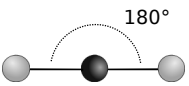
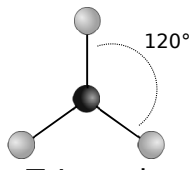
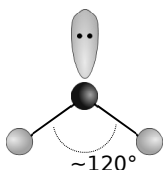
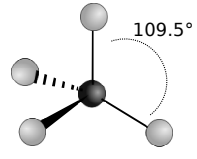
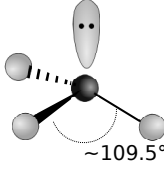
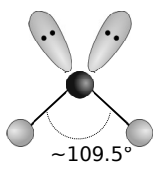
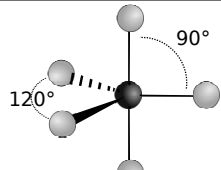
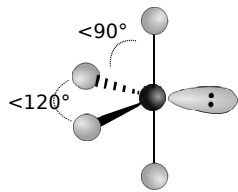
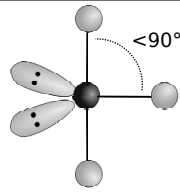
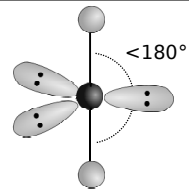
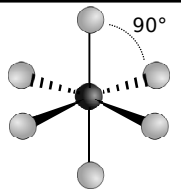
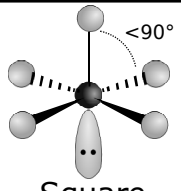
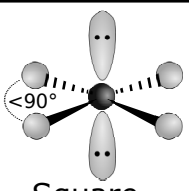
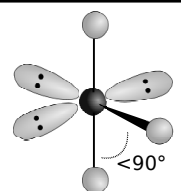
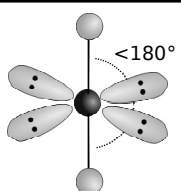


Valence Shell Electron Pair Repulsion (VSEPR) Geometries

Steric #	(Base Geometry)				
	0 Lone Pairs	1 Lone Pair	2 Lone Pairs	3 Lone Pairs	4 Lone Pairs
2	 Linear				
3	 Trigonal Planar	 Bent			
4	 Tetrahedral	 Trigonal Planar	 Bent		
5	 Trigonal Bipyramid	 See-Saw	 T-Shaped	 Linear	
6	 Octahedral	 Square Pyramid	 Square Planar	 T-Shaped	 T-Shaped

Note: Any structure with only two bonded atoms only is linear.