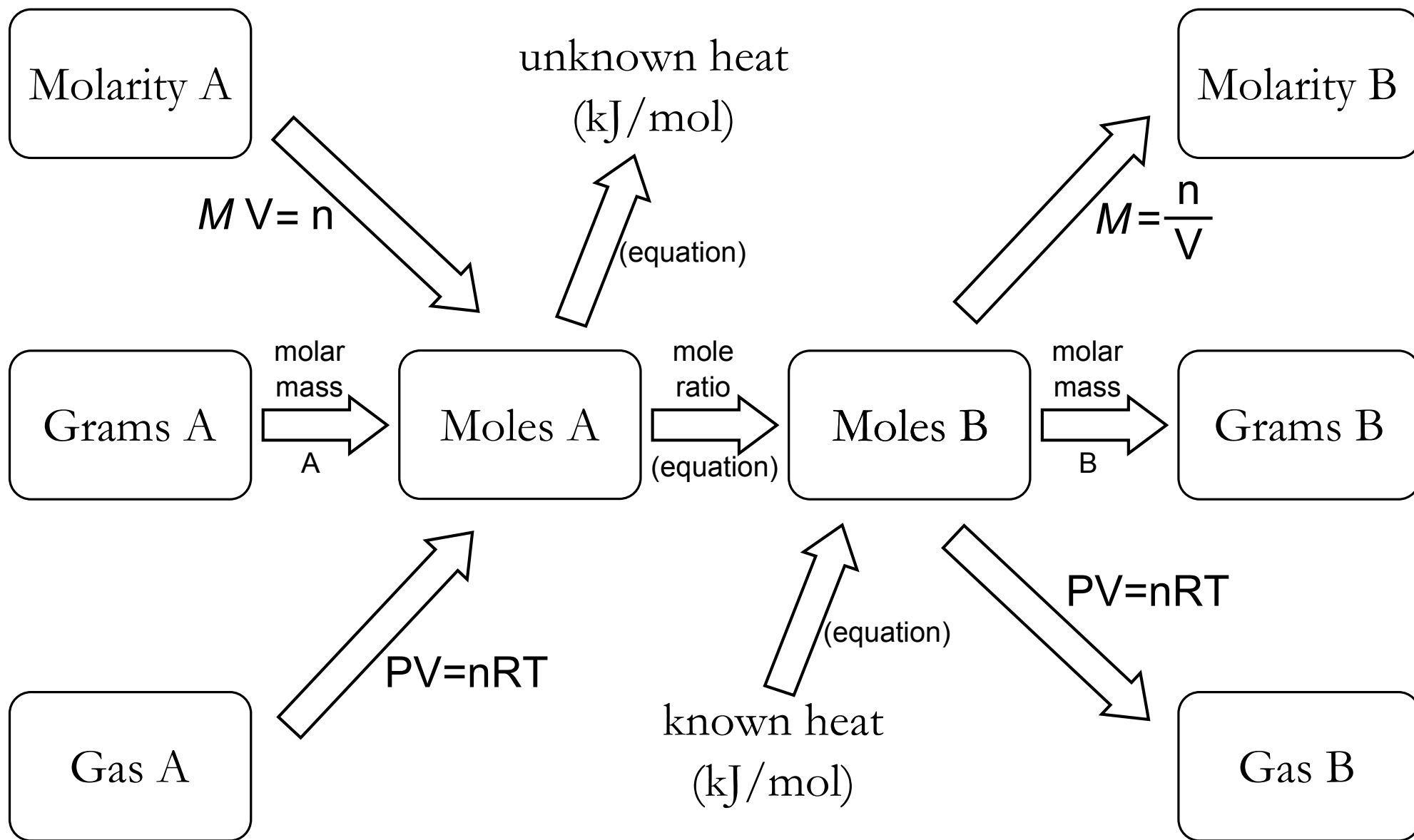


Stoichiometry Chart



Stoichiometry Notes

- “A” represents the known quantity of substance A, which can be a reactant or product.
- “B” represents an unknown quantity of substance B, which can be a reactant or product.
- If heat is provided as a known, we enter the equation with heat, to calculate moles B, which then may be converted to the desired (molarity,grams,gas property)
- If heat is unknown, then we obtain it from known Moles of A, which is either provided, or must be obtained by converting from (molarity,grams,gas properties).